

ELECTRON SPIN-LATTICE RELAXATION OF Cu(II) IN Zn(II)-BIS(DIETHYL-DISELENOCARBAMATE) SINGLE CRYSTALS

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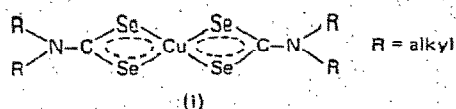
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Using the pulse saturation method the spin-lattice relaxation rate T_1^{-1} for Cu(II) in Zn(II)-bis(diethyl-diselenocarbamate) was measured in the temperature range $1.5 < T < 33$ K. From the T -dependence of T_1^{-1} the Debye temperature of the host crystal was obtained. The angular dependence of T_1^{-1} as well as the influence of the hyperfine interaction on T_1^{-1} are discussed.

1. Introduction

EPR investigations on single crystals and powders of Cu(II)-bis(dialkyl-diselenocarbamates) $\text{Cu}(\text{desc})_2(\text{I})$ diamagnetically diluted in various host lattices [1-5] showed some interesting features: (a) very small line widths of the Cu hfs lines, (b) non-coinciding principal axes of the g - and the $^{63,65}\text{Cu}$ -hfs tensors and (c) a very high degree of covalency in the Cu-Se bonds.



In this paper we report detailed investigations of the electron spin-lattice relaxation in single crystals of $\text{Cu}(\text{desc})_2$, diamagnetically diluted by the corresponding Zn(II) complex, in order to explain the line width behaviour. Possible influences of the covalency on the spin-lattice relaxation will also be discussed qualitatively. Until now there has been no information on the spin-lattice relaxation of ions of the iron group in crystals with very covalent metal-ligand bonds except for an investigation of the phonon bot-

tleneck effect in Cu(II)-maleonitril-dithiolate single crystals [6].

2. Experimental

$\text{Cu}(\text{desc})_2$ and $\text{Zn}(\text{desc})_2$ were obtained as described earlier [7,8]. Single crystals containing about 1.5% of the copper complex were grown by slow evaporation of a chloroformic solution of both complexes. Both $\text{Zn}(\text{desc})_2$ and $\text{Cu}(\text{desc})_2$ form monoclinic crystals, space group $P 2_1/c$, with four molecules in the unit cell [9]. In each case two of these are centro-symmetrically related, forming dimeric units. The local symmetry on the Zn atom can be considered as a distorted tetrahedron, that of the Cu atom is pyramidal. As suggested by Keijzers [5], in the diluted state the copper complex does not accept the structure of the pure Zn-complex, but has a structure which resembles strongly that of the pure $\text{Cu}(\text{desc})_2$.

Single crystal EPR spectra were recorded on an X-band spectrometer THN-251 (Thomson, France) and on a superheterodyne spectrometer at $\nu = 28.9$ GHz in the temperature range 4.2–295 K. The magnetic field was controlled by an NMR-marker; polycrystalline DPPH was used as the g -marker.